

Purity, Precision and Performance Redefined

Scientific community globally rely on pure water for a myriad of purposes within laboratories. Utilizing the appropriate level of water purity is a fundamental strategy to attain more consistent and accurate results in laboratory work. By ensuring a reliable source of pure water within the laboratory, scientists can enhance the reproducibility of experiments, minimize the risk of contamination, and optimize the overall efficiency of their research processes. The adoption of in-house purification systems aligns withsustainable practices, reflecting a commitment to environmental responsibility within the scientific communit

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Water Impurities & Source

Water is often referred to as the "Universal solvent" due to its excellent solvent properties and contains common impurities

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Dissolved Minerals like calcium, magnesium, sodium and potassium

Suspended Solid particles, such as sand, silt, and clay

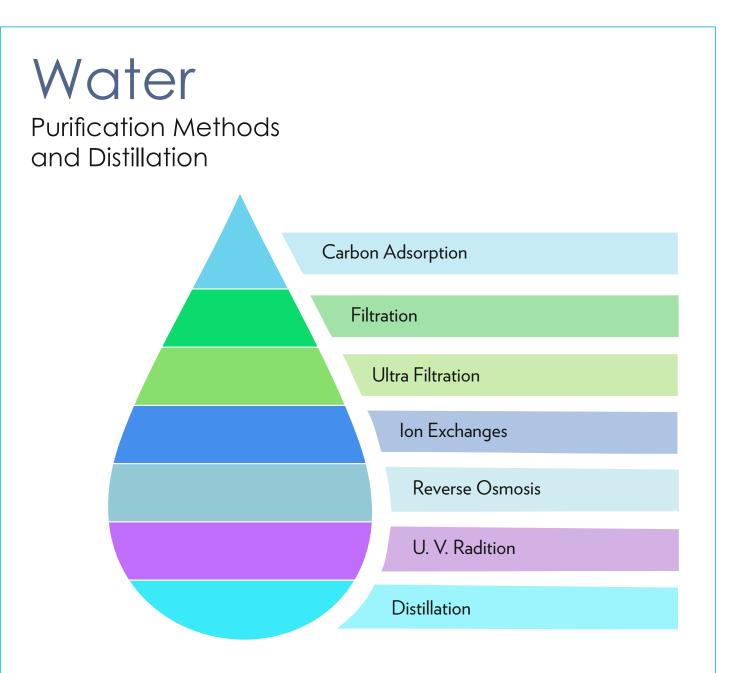
Microorganisms: Bacteria, viruses, algae, and other pyrogens

various chemicals, including pesticides, fertilizers, heavy metals, and organic pollutants

Natural organic matter, such as decaying plant and animal material

Dissolved Gases viz. Oxygen, Nitrogen and carbon dioxide,

It is important to note that water source exhibits significant variations from one geographical location to another, and these fluctuations can extend to changes from season to season and environmental conditions.



Of all above Distillation is probably the oldest method of water purification, Involving heating water to create steam, which is then condensed and collected back into liquid form, leaving impurities behind in boiling vessel.

Distillation remains most environmentally sustainable and economical while effectively removing wide range of impurities, including minerals, heavy metals, and volatile organic compounds, Pathogen Elimination by killing bacteria, viruses, and other microorganisms, thus producing high-purity water with low conductivity, making it suitable for various applications, including laboratory work, medical uses, and in industries requiring ultrapure water.

Standard for reagent grade water highlights the superior water quality of water achieved through Quartz or Vitreous Silica stills over water obtained using metal stills. Combining different methods like Distillation, Deionization, Reverse osmosis is often done to achieve comprehensive purification to achieve the specific water quality requirements and the intended use of the water.

All Quartz Water Distillation Units from Borosil facilitate automatic on-demand operation, promotes safe working environment by effectively purifying water through the distillation process.

The Borosil Scientific Quartz Cabinet Double Distillation Unit (QCD) is crafted from Pure Quartz, boasting a remarkable 99.95% SiO2 purity. Quartz, with impurities at a PPM level, is exceptionally well-suited for water distillation, ensuring the highest level of purity.





Quartz Cabinet Distillation (QCD250)



All Quartz Construction 99.5% SiO²



Cabinet Assembly Promotes Space-Efficient and Safe Usage



On Demand Output Produces desired amount of water and auto shut down



Safe Cut-offs/ Dry Run Cut-Off Safety of user and instrument, along with longevity of the instrument



Water Reservoir Distilled water is stored eliminating contamination



Point of Use Dispensing Distilled water in small quantities dispensed at your fingertips, for daily lab applications



Optical Level Sensor Monitors Reservoir filling and a uto on/off of unit.

Quartz Cabinet Double Distillation Unit

Technical Information

Model	Quartz Cabinet Distillation (QCD)
Capacity Distilled Water Output	2.5 LPH
Minimum cooling water requirement (Ltr/min)	2
Heater	Quartz
Boiler	Quartz
Condenser	Quartz
Biological Activity	Pyrogen Free
*Conductivity S/cm	<1 x 10-6
Power consumption (kw)	4.4 kW
Voltage (V)	230V
Dry Run Protection	Yes
On Demand Output	Yes
Storage Mode	Yes
Auto Filling Reservoir	Yes
Dispense rate (ml/min)	900 ml
HSN Code	84194090
Product Code	BLFRQCD2500000000

*1 Inlet water TDS < 100 ppm

Note: Because of difficulties associated with measurement of the pH value of high purity water and the doubtful significance values obtained, limits for the pH value of grade 1 & grade 2 water have not been specified.

Pure water is an excellent solvent and will dissolve carbon dioxide from the atmosphere to form a very dilute solution of carbonic acid with a pH below 7. Pure water typically exhibits a conductivity of $1-2 \mu$ S/cm and a pH of 5, primarily due to the slight absorption of CO2 (0.5 ppm) from the surrounding environment.

Laboratory Applications of **Distilled Water**

Agriculture Research & Testing



Laboratory Research involvingchemical analysis & biological research

Water Testing



Quality Testings involving chemical analysis & biological research

Environment Testing



Prepare standards, calibrate instruments, buffer preparations

Microbiology



Media Preparations, Buffers, Reagent Dilutions, dilute stains and fixatives, Buffered Peptone Water Preparation

Dairy Industry



Media Preparations, Buffers, Reagent Dilutions, Bacterial Plate counting, Somatic Cell Counting, Residue tests for possible contaminations.

Cosmetics



Emulsion Formulation, Hydration toskin care,



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